

- (21) Application No. 15252/78 (22) Filed 23 Mar. 1976
(61) Divided out of No. 1528341
(31) Convention Application No. 562648 (32) Filed 27 Mar. 1975 in
(33) United States of America (US)
(44) Complete Specification Published 11 Oct. 1978
(51) INT. CL.² C08J 3/24
(52) Index at Acceptance
C3P D9A1A D9D10
(72) Inventors: David Apotheker
Paul Joseph Krusic

(19)



(54) BROMINE-CONTAINING FLUOROPOLYMER COMPOSITION

(71) We, E.I. DU PONT DE NEMOURS AND COMPANY, a corporation organized and existing under the laws of the State of Delaware, located at Wilmington, State of Delaware, United States of America, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:-

This invention relates to fluoropolymer compositions which are useful in the manufacture of cured fluoropolymer articles.

According to the invention, we provide a fluoropolymer composition which comprises (A) a copolymer which includes interpolymers of units of

(1) up to 3 mole % of units derived from a bromine-containing olefin with the proviso that enough of such units are present to provide at least 0.05 weight % bromine in the copolymer, and

(2) enough units derived from component (2.1) or (2.2) as follows to make up the remainder of the copolymer:

(2.1) at least one fluorine-containing monomer which is copolymerizable with component (1), contains 2-7 carbon atoms, contains no bromine atom, and contains at least as many fluorine atoms as carbon atoms, or

(2.2) 50-99 mole % of tetrafluoroethylene, 1-50 mole % of a C₂-C₄ olefinic hydrocarbon and from 0 to 30 mole % of vinylidene fluoride, and

(B) 0.5-10% of an organic peroxide based on the weight of component (A),

The copolymer (A) employed in the compositions according to the invention is preferably a fluoropolymer which includes interpolymers of units of

(a) up to 3 mole %, based on the total moles of components (a) and (b), of units derived from bromotrifluoroethylene or 4-bromo-3,3,4,4-tetrafluorobutene-1, with the proviso that enough of such units are present to provide at least 0.05 weight % bromine in the copolymer, and

(b) enough units derived from alternative components (b-1), (b-2) or (b-3) as follows to make up the remainder of the copolymer:

(b-1) tetrafluoroethylene and a perfluoroalkyl perfluorovinyl ether wherein the perfluoroalkyl group contains 1-5 carbon atoms,

(b-2) 45-65 mole % tetrafluoroethylene, 20-55 mole % of a C₂-C₄ olefinic hydrocarbon and from 0 to 30 mole % vinylidene fluoride, or

(b-3) vinylidene fluoride, at least one fluorine-containing olefin copolymerizable therewith, and optionally said perfluoroalkyl perfluorovinyl ether, said olefin being a compound which contains 2-7 carbon atoms, contains no bromine atom, and contains at least as many fluorine atoms as carbon atoms.

Such fluoropolymers are described and claimed in our copending Application No. 11658/76 (Patent No. 1,528,341).

In general, the bromine-containing olefin used to form the units of component (1) in the copolymer of this composition is preferably bromotrifluoroethylene or 4-bromo-3,3,4,4-tetrafluorobutene-1, or a blend of these two compounds. The term "bromine-containing olefin" herein means an olefin in which at least one hydrogen atom has been replaced with a bromine atom, and optionally on or more of the remaining hydrogen atoms have been replaced with an atom of another halogen, preferably fluorine. Some compounds of this type are available commercially and others can be prepared by method known in the art, for

example as shown by Tarrant and Tunden in 34 J. Org. Chem. 864 (1969) and by Fainberg and Miller in 79 JACS 4170 (1957). Other bromine-containing olefins besides the two already mentioned which are copolymerizable with the monomers used to form the component (2) units of copolymer, and useful to form the component (A-1) units, are illustrated as follows:

vinyl bromide, 1-bromo-2,2-difluoroethylene, perfluoroalkyl bromide, 4-bromo-1,1,2-trifluorobutene, 4-bromo-1,1,3,3,4,4-hexafluorobutene, 4-bromo-3-chloro-1,1,3,4,4-pentafluorobutene, 6-bromo-5,5,6,6-tetrafluorohexene, 4-bromoperfluorobutene-1, and 3,3-difluoroalkyl bromide.

It is usually preferred that enough of the component (A-1) units are present to provide 0.3-1.5 weight % bromine in the copolymer.

As shown above, the component (2) units of the copolymer (A) used in the composition can be derived from a specified type of fluorine-containing monomer (referred to for the sake of convenience as component 2.1); or the component (2) units can be derived from a specified blend (referred to as component 2.2) of tetrafluoroethylene, an olefinic hydrocarbon and optionally vinylidene fluoride.

In a preferred embodiment of the composition, component 2.1 is composed of units derived from tetrafluoroethylene and perfluoroalkyl perfluorovinyl ethers wherein the perfluoroalkyl group contains 1-5 carbon atoms. Said ether is preferably perfluoromethyl perfluorovinyl ether. In some of the best illustrations of this type of composition, the fluoropolymer is a fluoroelastomer wherein the tetrafluoroethylene content of component 2.1 is 50-80 mole % and the ether content is 20-50 mole %.

In another preferred embodiment of the composition, component 2.1 is composed of units derived from vinylidene fluoride, at least one fluorine-containing olefin copolymerizable therewith, and optionally a perfluoroalkyl perfluorovinyl ether wherein the alkyl group contains 1-5 carbon atoms. It is especially preferred in this type of composition that the fluoropolymer is a fluoroelastomer wherein component 2.1 is composed of units derived from any one of the following three combinations:

(i) vinylidene fluoride and hexafluoropropylene or pentafluoropropylene; or

(ii) vinylidene fluoride, tetrafluoroethylene and hexafluoropropylene or pentafluoropropylene; or

(iii) vinylidene fluoride, a perfluoroalkyl perfluorovinyl ether wherein the perfluoroalkyl group contains 1-5 carbon atoms, and hexafluoropropylene and/or tetrafluoroethylene.

For example, component 2.1 of one such fluoroelastomer is composed of 30-70 weight % vinylidene fluoride units and 30-70 weight % of hexafluoropropylene units. Another such fluoroelastomer has component 2.1 composed of 25-70 weight % vinylidene fluoride units 19-60 weight % hexafluoropropylene units and 3-35 weight % tetrafluoroethylene units. Monomer combinations useful as component 2.1 in other such fluoroelastomers are described above in the discussion of component (b-3) of the novel fluoropolymer. Component 2.1 can also be composed of tetrafluoroethylene alone or vinylidene fluoride alone in applications wherein it is not necessary that the fluoropolymer be a fluoroelastomer.

As already mentioned, the component (2) units of the copolymer of the present composition can also be derived from the specified blend (tetrafluoroethylene, olefinic hydrocarbon and optionally vinylidene fluoride) referred to as component (2.2). For example, component 2.2 can be composed of (units derived from) 50-99 mole % tetrafluoroethylene and 1-50 mole % propylene. Component 2.2 can also be composed of 1-30 mole % vinylidene fluoride units, 50-98 mole % tetrafluoroethylene and 1-48 mole % propylene.

Component B of the present composition is an organic peroxide, preferably a dialkyl peroxide. An organic peroxide is selected which will function as a curing agent for the composition in the presence of the other ingredients which are to be present in the end-use composition and under the temperatures to be used in the curing operation without causing any harmful amount of curing during mixing or other operations which are to precede the curing operation. A dialkyl peroxide which decomposes at a temperature above 49°C is especially preferred when the composition is to be subjected to processing at elevated temperatures before it is cured. In many cases one will prefer to use a di-tertiarybutyl peroxide having a tertiary carbon atom attached to a peroxy oxygen. Among the most useful peroxides of this type are 2,5-dimethyl-2,5-di(tertiarybutylperoxy) hexyne-3 and 2,5-dimethyl-2,5-di(tertiarybutylperoxy) hexane. Component B can also be selected from such compounds as dicumyl peroxide dibenzoyl peroxide, tertiary butyl perbenzoate and di[1,3-dimethyl-3-(t-butylperoxy)butyl]carbonate.

To illustrate an application where one can use a component B compound which decomposes at or below 49°C, one can carry out the steps of preparing the composition at room temperature as a liquid film-forming composition containing an organic solvent and/or diluent, applying a layer of the liquid composition to a substrate, and allowing the layer to become dried and cured at or below 49°C.

One will usually prefer to employ a component B peroxide content of 1.5-5% based on the weight of the component A copolymer.

The present composition in the form of a blend of the component A copolymer and the component B peroxide is a useful article of commerce which (as in the case of a novel polymer) can be supplied to manufacturers of fabricated fluoropolymer articles who wish to modify the material with certain additives according to the requirements of particular applications before it is molded, extruded or otherwise made into cured end-products. Another useful approach is for the supplier of the composition to add one or both of components C and D described below before the composition is delivered to the person who makes the cured products.

One material which is usually blended with the composition before it is made into end products is at least one metal compound selected from divalent metal oxides and divalent metal hydroxides. This material can be referred to as component (C). A metal salt of a weak acid can be used along with the oxide and/or hydroxide. Component (C) will generally be added in an amount equal to 1-15%, preferably 2-10%, by weight of the copolymer content. Metal compounds useful as component (C) are further described by Bowman in U.S. Patent 3,686,143.

Another material which is usually blended with the composition before it is made into end products is a co-agent composed of a polyunsaturated compound which is capable of cooperating with said peroxide to provide a useful cure. This material, which can be referred to as component (D), will generally be added in an amount equal to 0.5-10% preferably 1-7%, by weight of the copolymer, content. It is preferred in many applications to use as a component (D) co-agent one or more of the following compounds: triallyl cyanurate; triallyl isocyanurate; tris (diallylamine)-s-triazine; triallyl phosphite, N,N-diallyl acrylamide; hexa-allyl phosphoramidate; N,N,N',N'-tetra-allyl terephthalamide; N,N,N',N'-tetra-allyl malonamide; trivinyl isocyanurate; 2,4,6-trivinyl methyltrisiloxane; and tri(5-norbornene-2-methylene)cyanurate; and the like. Particularly useful is triallyl isocyanurate.

The composition can also contain one or more additives such as those known to be useful in fluoropolymer compositions, for example pigments, fillers, pore-forming agents and liquid organic solvents. An example of a useful type of solvent is the one described by Proskow in U.S. Patent 3,740,369. Others include acetone, methyl ethyl ketone, ethyl acetate, amyl acetate and, tetrahydrofuran. By adding a suitable solvent, one can prepare a liquid composition useful in the preparation of adhesive layers, coatings and films.

In preparing the present fluoropolymer composition, one can mix the copolymer with the other ingredients by means of any mixing apparatus known to be useful for preparing rubber or plastic compositions; for example, one can use a roller-type rubber mill or Banbury mixer equipped to operate at a temperature below the decomposition temperature of the organic peroxide.

The composition can be cured by subjecting it to conditions which result in the decomposition of the organic peroxide, for example by heating the composition at a temperature which causes the peroxide to decompose. The initial curing of the curable composition containing components A, B, C and D in most cases is preferably carried out by heating the composition for 1-60 minutes at 149-204°C; conventional rubber- and plastic-curing presses, molds, extruders, provided with suitable heating and curing means can be used. Also, if one wants a product having maximum heat resistance and dimensional stability, it is preferred to carry out a post-curing operation wherein the article is heated in an oven or the like for an additional period of 1-48 hours at 180-300°C. One skilled in the art will realize that the best curing time and temperature for a particular application will depend on such factors as the nature and proportion of ingredients and the properties needed in the final product.

Fluoropolymer plastics and elastomers, and compositions containing them, can easily be made in accordance with the present invention in vulcanizable grades suitable for compounding and curing by practical and economical methods to yield highly useful cured fluoropolymer plastic and elastomer articles for applications such as films, gaskets, O-rings, coated fabrics, wire insulation, hoses and protective coatings, which have good creep-resistance and good resistance to damage by heat, solvents and corrosive chemicals. Moreover, one can prepare fluoropolymers for use in the present composition from a wide variety of fluorine-containing monomers to obtain properties in the cured plastic and elastomer articles suited for a wide variety of applications.

To illustrate the beneficial utility of the present products further, fluoropolymers are obtainable within the present invention: (a) which can be cured rapidly; (b) which can be cured at atmospheric pressure to a non-porous or non-spongy structure, an important advantage in the manufacture of fluoropolymer-coated wire and fluoropolymer hose and the like; (c) which can be cured at about ordinary room temperatures; (d) which undergo a surprisingly great increase in torque during the ODR cure test described below, which is

desirable in many applications; (e) which have lower (i.e. more useful) compression set values after curing than prior art diamine-cured fluoroelastomers (this applies to certain fluoroelastomer compositions of the present invention wherein the fluoroelastomer is a vinylidene fluoride copolymer) and (f) which can be cured with less tendency to undergo polymer degradation than certain prior art fluoroelastomer materials cured with a blend of aromatic hydroxy compound and quaternary phosphonium compound of the type shown by Patel et al in U.S. 3,712,877. (This applies to certain fluoroelastomer compositions of this invention wherein the fluoroelastomer is a vinylidene fluoride copolymer containing units derived from a perfluoralkyl perfluorovinyl ether.)

The following examples illustrate the invention; all amounts are by weight unless otherwise indicated. Inherent viscosities quoted in the Examples are expressed as deciliters/gram (dl/g).

Example 1

A curable fluoropolymer in the form of an elastomeric copolymer of tetrafluoroethylene, perfluoromethyl perfluorovinyl ether and bromotrifluoroethylene is prepared by a continuous process composed of the following operations:

(1) Continuously feeding the three monomers described below in Table I to a 3.8 liter stainless steel pressure vessel reactor (i.e. the polymerization reaction zone) which has been flushed with nitrogen, while operating the stirrer of the reactor at 500 rpm for thorough mixing of the reactor contents, and while the contents of the reactor are heated at 70°C, under a pressure of 42 kg./cm.² so that the reaction mixture formed in operation (2) below will undergo an emulsion polymerization reaction as it passes through the reactor, the reactor residence time being about 2.7 hours based on the ratio of the 3.8 liter reactor to the emulsion output rate of about 1.4 liters per hour, the monomers and the feed rate for each being as shown below in Table I;

(2) During operation 1, constantly feeding to the reactor through a first metering pump during each hour a solution composed of 6.38 grams of ammonium persulfate, 4.5 grams of dibasic sodium phosphate heptahydrate and 12.0 grams of ammonium perfluoro octanoate dissolved in 600 ml. of water (distilled), and simultaneously feeding to the reactor through a second metering pump during each hour a solution composed of 5.25 grams of sodium sulfite dissolved in 600 ml. of water, the reaction mixture being maintained at a pH of 6.1; (3) Continuously removing from the reactor the resulting copolymer latex which is continuously formed during operations 1 and 2, the latex being passed first through a back-pressure regulating valve set to maintain the desired reactor pressure of 42 kg./cm.², and then through a sealed container from which the off-gas (unreacted monomers) is led to a gas chromatograph where its composition is determined and recorded in Table I below;

(4) After discarding the latex obtained during the first four residence times, collecting the desired quantity of latex and mixing it for uniformity, the latex having a pH of about 6.1 and a copolymer solids content of 26.0%; and

(5) isolating the resulting copolymer from the latex by the gradual addition of a 4% aqueous solution of potassium aluminum sulfate until the copolymer is coagulated, washing the copolymer particles with distilled water, removing the water by means of a filter apparatus, and then drying the copolymer in a circulating air-oven at 100°C to a moisture content of less than 1%.

Table I

Monomer	Feed g./hr.	Off-Gas g./hr.	Incorporated in Polymer		
			g./hr.	Wt. %	Mole %
Tetrafluoroethylene	260	14	246	56.0	68.5
Perfluoromethyl perfluorovinyl ether	300	117	183	41.6	30.0
Bromotrifluoroethylene	10	1	9	2.4	1.5
Total	570	132	438		

The resulting copolymer is an elastomer which can be cured in the manner described below in Example 2; as indicated in Table I above, it contains in each 100 grams of polymerized units about 56 grams of polymerized tetrafluoroethylene, 41.6 grams of polymerized perfluoromethyl perfluorovinyl ether and 2.4 grams of bromotrifluoroethylene. The copolymer has an inherent viscosity of 0.38; this property is measured at 30°C at a polymer concentration of 0.2% by weight in a solvent composed of 58% by volume of

2,3,3-trichloroperfluorobutane, 39% by volume of "FC-75" solvent and 3% by volume of diethylene glycol dimethyl ether. The "FC-75" solvent is a perfluoro cyclic ether type of solvent of the empirical formula $C_8F_{16}O$, sold under the 3M Company trade name "FC-75", and described in further detail in the above-mentioned Proskow U.S. Patent 3,740,369. In place of this solvent one can use the product of fluorinating tetrahydrofuran until one obtains a solvent composed predominantly of 2-perfluorobutyl perfluorotetrahydrofuran. The copolymer has a bromine content of 1.2% by weight; and it is insoluble in aliphatic and aromatic hydrocarbons. The copolymer can be blended with additives in the manner described in Example 2 and used in the manufacture of O-rings and other cured elastomer articles.

Example 2

A curable fluoroelastomer is prepared by repeating Example 1 except the monomer feed rate is adjusted so that a copolymer is obtained which contains 65.0 mole % of tetrafluoroethylene units, 33.5 mole % of perfluoromethyl perfluorovinyl ether units and 1.5 mole % of bromotrifluoroethylene units. The resulting Example 2 copolymer has the same bromine content and inherent viscosity as the Example 1 copolymer.

A curable fluoroelastomer composition is prepared by mixing the following ingredients on a two-roll rubber mill whose rolls are at about 60°C: 100 parts of the Example 2 copolymer, 10 parts of carbon black (SAF), 4 parts of magnesium oxide, 2 parts of calcium oxide, 4 parts of triallyl isocyanurate and 4 parts of peroxide curing agent (45% of 2,5-dimethyl-2,5-di(tertiarybutylperoxy)hexyne-3 and 55% inert filler, sold by Wallace and Tiernan as "Luperco 130 XL". Luperco is a registered Trade Mark.

A sample of the resulting composition is used for measuring curing characteristics by means of the ODR (oscillating disc rheometer) Cure Test described by Pattison in U.S. 3,876,654 at a test temperature of 177°C. At a curing time of 30 minutes, the amount of torque increase is 138 cm. kg. This relatively great torque increase is unexpected and beneficially useful in many applications compared with that obtained in some of the more useful prior art curable fluoroelastomer compositions; a very rapid and high state of cure is indicated here.

The Shore Hardness (Durometer A) value of the composition is 92 when tested by ASTM Method D-676 on samples of the composition press-cured for 15 minutes at 190°C under a total pressure of 18150 kg., and post cured in an oven during a cycle of: 6 hours to reach 204°C, 18 hours at 204°C, 6 hours to reach 288°C and 18 hours at 288°C.

The compression set value of the composition is 50 when tested at 204°C for 70 hours by ASTM Method D-395-61 Method B, using as test samples pellets (1.27 cm. thick and 1.9 cm. in diameter) cured as described in the prior paragraph.

Example 3

A curable fluoroelastomer is prepared, compounded and cured in the manner described in Example 2 except:

(a) the monomer feed rate is adjusted so that a copolymer is obtained which contains 65.6 mole % tetrafluoroethylene units, 33.0 mole % perfluoromethyl perfluorovinyl ether units and 1.4 mole % bromotrifluoroethylene units. This copolymer has a bromine content of 0.9 weight % and an inherent viscosity (Example 1 method) of 0.35;

(b) in preparing the curable composition, the amount of carbon black is 5 parts, the amount of magnesium oxide is 5 parts and the amount of peroxide curing agent is 5 parts; and

(c) test samples are press-cured for 30 minutes at 180°C and post-cured as in Example 2.

The compression set value of the cured Example 3 fluoroelastomer is 57 when tested as in Example 2. Tensile strength is 147 kg./cm.², elongation at break is 60% and set at break is 3% (ASTM D-412).

Example 3A

For purposes of comparison, a fluoroelastomer outside the invention is prepared, compounded and cured in the manner described in Example 3 except the monomer feed rate is adjusted (the amount of bromotrifluoroethylene is increased) so that a copolymer is obtained whose bromine content is 7.2 weight % (equal to 11.4 mole % bromotrifluoroethylene units). The tetrafluoroethylene/ether ratio is kept at about 2/1 as in Ex. 3.

The cured Example 3A test samples are so brittle that one cannot obtain values for tensile strength, elongation and set at break.

Example 4

A curable fluoroelastomer composition is prepared and cured in the manner described in Example 2 except the bromotrifluoroethylene is replaced with bromodifluoroethylene; and the monomer feed rate is adjusted to obtain a copolymer which contains 69.63 mole % tetrafluoroethylene units, 30.0 mole % perfluoromethyl perfluorovinyl ether units and 0.37 mole % bromodifluoroethylene units. This copolymer has a bromine content of 0.3 weight % and an inherent viscosity (Example 1 method) of 0.27.

In the ODR (oscillating disc rheometer) Cure Test described in Example 2, the torque increase is 46 cm. kg. at a curing time of 30 minutes. The composition has a Shore A Hardness of 90, a compression set of 99 (Example 2 method), a tensile strength at break (T_B) of 194 kg./cm.², an elongation at break (E_B) of 140%, a modulus at 100% elongation (M₁₀₀ per ASTM D-412) of 147 kg./cm.² and a set at break of 19%.

Example 5

A fluoroelastomer composition is prepared as described in Example 2 except the bromotrifluoroethylene is replaced with perfluoroallyl bromide; and the monomer feed rate is adjusted to obtain a copolymer which contains 69.2 mole % tetrafluoroethylene units, 30.0 mole % of perfluoromethyl perfluorovinyl ether units and 0.8 mole % of perfluoroallyl bromide units. This copolymer has a bromine content of 0.3 weight % and an inherent viscosity of 0.21.

In the ODR Cure Test, the torque increase at 30 minutes is 19.6 cm. kg. The composition has a T_B of 151 kg./cm.², an E_B of 100%, an M₁₀₀ of 151 kg./cm.² and a set at break of 13%.

Example 6

A fluoroelastomer composition is prepared as described in Example 2 except the bromotrifluoroethylene is replaced with 3,3-difluoroallyl bromide; and the monomer feed rate is adjusted to obtain a copolymer which contains 66 mole % of tetrafluoroethylene units, 33 mole % of perfluoromethyl perfluorovinyl ether units and 1 mole % of units of said bromide. The copolymer has a bromine content of 0.7 weight %.

In the ODR Cure Test, the torque increase at 30 minutes is 23 cm. kg. The composition has a T_B of 103 kg./cm.², an E_B of 90% and a set at break of 8%.

Example 7

A curable non-elastomeric fluoropolymer composition is prepared as follows: A copolymer which contains 97 mole % tetrafluoroethylene units and 3 mole % bromotrifluoroethylene units is made by (a) providing a 400 cc. shaker bomb containing 0.75 grams ammonium persulfate, 200 cc. of water, 0.15 gram of ammonium perfluoro-octanoate, 50 grams of tetrafluoroethylene and 3 grams of bromotrifluoroethylene; (b) heating the mixture in the bomb for 3 hours at 85°C; (c) cooling the mixture to 24° C; (d) venting the gas from the bomb; (e) using a Buchner funnel to filter the copolymer and wash it three times with water; and (f) drying the copolymer in a vacuum oven for 14 hours at 70°C.

The copolymer is a white powder having a bromine content of 2.6 weight % and a melting point of 315-320°C.

The curable composition is made by mixing the following in a ball-mill for three days at 25°C: 100 parts of the Example 7 copolymer, 4 parts of magnesium oxide, 2 parts of calcium oxide, 4 parts of triallyl isocyanurate and 5 parts of the peroxide curing agent of Example 2.

A sample of the resulting composition is compression molded while heated at 204°C for 1 hour. The resulting cured fluoropolymer plastic material does not melt when heated well above 320°C, and shows no evidence of melting when it is heated in an open flame.

Example 8

A fluoroelastomer is prepared by repeating Example 1 except: the monomers fed to the reactor in the present Example are 56 parts vinylidene fluoride, 44 parts hexafluoropropylene and 0.8 part bromotrifluoroethylene; a two-liter reactor is used; the pressure in the reactor is 63 kg./cm.²; the reaction mixture temperature is 105°C and the pH is 3.3; the reactor residence time is about 15 minutes; and operation (2) consists of feeding to the reactor (for each 100 parts of monomer) 400 parts water containing 0.3 part of ammonium persulfate and 0.05 part sodium hydroxide. The gas chromatograph of operation (3) can be omitted. The copolymer obtained, after 91% monomer conversion, contains 59.1% vinylidene fluoride units, 40% hexafluoropropylene units and 0.9% bromotrifluoroethylene units (weight %). The latex has a copolymer solids content of 18.7%.

The resulting fluoroelastomer has a Mooney viscosity of 94 at 100°C; this measure of bulk viscosity is determined on a Mooney viscometer using the large rotor and a ten-minute shearing time.

Example 9

A curable fluoroelastomer composition is prepared by mixing the following on a two-roll rubber mill whose rolls are at about 25°C: 100 parts of the Example 8 copolymer, 15 parts of carbon black MT, 10 parts of finely pulverized bituminous coal ("Austin Black"), 1 part of calcium hydroxide, 2 parts calcium silicate ("Microcel E"), 4 parts of triallyl isocyanurate, and 4 parts of peroxide curing agent ("Luperco 101XL", which is understood to be 45% of 2,5-dimethyl-2,5-di(tertiarybutylperoxy) hexane and 55% inert filler).

Test samples of the composition are press-cured for 30 minutes at 177°C and post-cured in an oven during a cycle of 4 hours to reach 260°C and 18 hours at 260°C.

The resulting cured fluoroelastomer has a compression set value of 27 (Example 2 method), a T_B of 154 kg./cm.² an E_B of 170% and an M₁₀₀ of 75.5 kg./cm.².

Example 10

Examples 8 and 9 are repeated except for the changes mentioned below.

The copolymer is prepared from 82 parts vinylidene fluoride, 67 parts hexafluoropropylene, 1.4 parts 4-bromo-3,3,4,4-tetrafluorobutene-1, and (for each 150 parts monomer) 600 parts of water containing 0.4 part of ammonium persulfate and 0.08 part of sodium hydroxide. The reaction mixture pH is 4.3, residence time is 20 minutes and the copolymer solids content of the latex is 18%.

The copolymer obtained, after 86% monomer conversion, contains 58.9% vinylidene fluoride units, 40% hexafluoropropylene units and 1.1% units of the bromine compound (weight %). The resulting fluoroelastomer has a Mooney viscosity of 100 (Example 8 method); and it has an inherent viscosity of 1.57 when measured at 30°C at a concentration of 0.1% by weight in a solvent composed of 87% by volume of tetrahydrofuran and 13% by volume of N,N-dimethyl formamide.

The curable fluoroelastomer composition is prepared by mixing on a rubber mill 100 parts of the Example 10 copolymer, 15 parts carbon black MT, 10 parts "Austin Black", 1 part magnesium oxide, 3 parts dibasic lead phosphite, 3.3 parts of triallyl isocyanurate and 3.3 parts of the Example 9 peroxide curing agent.

The test samples, when cured as in Example 9, have a compression set value of 25, a T_B of 144 kg./cm.², an E_B of 185% and an M_{100} of 59.6 kg./cm.².

When a fluoropolymer outside the invention is made, for purposes of comparison, in the substantially the same manner as in Example 10 except for the omission of the bromine compound, and the resulting copolymer of vinylidene fluoride and hexafluoropropylene is compounded and press-cured as in Example 10, a spongy non-useful product is obtained of the type described below in Example 13A. "Substantially the same" as used herein means the procedure used is the same for all practical purposes.

Example 11

A curable non-elastomeric fluoropolymer composition is prepared as follows: a copolymer which contains about 98% vinylidene fluoride units, 1.7% bromotrifluoroethylene units and a small amount of polymerized hexafluoropropylene units is made by repeating Example 8 except the copolymer is prepared by feeding to the reactor 100 parts vinylidene fluoride, 35 parts hexafluoropropylene, 1.6 parts bromotrifluoroethylene and (for each 100 parts vinylidene fluoride) 454 parts water containing 0.45 part ammonium persulfate, 0.09 part sodium hydroxide and 0.94 part ammonium perfluoro octanoate; the reactor pressure is 42 kg./cm.²; residence time is 30 minutes; the hexafluoropropylene feed is discontinued after polymerization is well established and the reaction is allowed to continue in the absence of hexafluoropropylene for 75 minutes. The copolymer is dried in a vacuum oven for 14 hours at 70°C. It has a bromine content of 0.85 weight %.

The curable composition is made and cured in the manner described in Example 7 except for the use of the Example 11 copolymer instead of the Example 7 copolymer. The resulting cured fluoropolymer plastic material is much more resistant to being dissolved by N,N-dimethyl acetamide than polyvinylidene fluoride when immersed in the solvent for several hours at 24°C.

Example 12

A fluoropolymer which is a rubbery plastic is prepared by repeating Example 8 except the copolymer is made from 69 parts vinylidene fluoride, 11 parts hexafluoropropylene, 19 parts tetrafluoroethylene and 1 part bromotrifluoroethylene, and (for each 100 parts of monomer) 400 parts water containing 0.85 part ammonium persulfate and 0.21 part sodium hydroxide. The reaction mixture pH is 4.6, residence time is 15 minutes and the latex copolymer solids content (after 96% monomer conversion) is 19.7%. The copolymer contains 70% vinylidene fluoride, 9% hexafluoropropylene, 20% tetrafluoroethylene and 1% bromotrifluoroethylene (weight %). The portion of the copolymer which dissolves (80%) has an inherent viscosity of 0.46 (method of Example 10).

A curable fluoropolymer composition is prepared by mixing the following in a ball mill for two days at 24°C: 100 parts of the Example 12 copolymer in the form of a fine powder, 5 parts magnesium oxide, 5 parts of the peroxide curing agent of Example 2, and a mixture of 4 parts of triallyl isocyanurate and 2 parts of calcium silicate.

A cured test pellet is formed by pressing a sample of the resulting composition for 30 minutes at 204°C under enough pressure to form a pellet. When the pellet is immersed for three days at 24°C in acetonitrile, it swells some and undergoes a weight gain of 100%. When a pellet formed in the same manner from the Example 12 copolymer (no additives) is given the same test, it undergoes considerable disintegration and a weight gain of 350%.

Example 13

A fluoropolymer is prepared as described in Example 8 except the copolymer is made from 77 parts tetrafluoroethylene, 23 parts propylene, 1.1 part bromotrifluoroethylene and (for each 100 parts of monomer) 359 parts of water containing 2.87 parts of surfactant (a 33%

aqueous solution of sodium lauryl sulfate), 1.65 parts ammonium persulfate and 1.08 parts sodium hydroxide. The reaction conditions: pressure of 42 kg./cm.², temperature of 80°C, residence time of 3 hours and pH of 9.0. The latex copolymer solids content (after 87% monomer conversion) is 20.7%. The copolymer contains about 73.4% tetrafluoroethylene, 25.3% propylene and 1.3% bromotrifluoroethylene (by weight). It has a Mooney viscosity of 61 (Example 8 method).

A curable fluoropolymer composition is prepared by mixing the following on a two-roll rubber mill: 100 parts of the Example 13 copolymer, 30 parts carbon black MT, 4 parts magnesium oxide, 2 parts calcium oxide, 4 parts triallyl isocyanurate and 5 parts of the peroxide curing agent of Example 2.

Test samples of the composition are press-cured 30 minutes at 177°C and 24 hours at 130°C.

The resulting fluoropolymer has a compression set value of 50 (O-rings tested 70 hours at 232°C, ASTM D-395), a T_B of 63 kg./cm.², an E_B of 200% and an M₁₀₀ of 28 kg./cm.².

Example 13A

For purposes of comparison, a fluoropolymer outside the invention is made in substantially the same manner as in Example 13 except for the omission of the bromotrifluoroethylene; and the resulting copolymer of tetrafluoroethylene and propylene is compounded and press-cured as in Example 13. The samples removed from the press have a spongy structure. The composition has no utility for normal fluoropolymer applications. No meaningful test data of the type shown in Example 13 could be obtained on this material.

Example 14

A fluoropolymer is prepared as described in Example 13 except the copolymer is made from 68 parts tetrafluoroethylene, 18.3 parts vinylidene fluoride, 13.1 parts propylene, 0.67 part bromotrifluoroethylene and (per 100 parts of monomer) 343 parts water containing 3.1 parts of the Example 13 surfactant, 1.96 parts ammonium persulfate and 1.37 parts sodium hydroxide. Residence time is 90 minutes. The latex copolymer solids content (after 67% monomer conversion) is 17.8%. The copolymer contains about 68.6% tetrafluoroethylene, 12.9% vinylidene fluoride, 17.5% and 1.0% bromotrifluoroethylene.

A curable fluoropolymer composition is prepared in the manner described in Example 13 except the Example 14 copolymer is used. Test samples are cured as described in Example 13 except the post-cure temperature is 204°C.

The resulting cured fluoropolymer has a compression set value of 85 (Example 13 method), a T_B of 80.4 kg./cm.², an E_B of 140% and an M₁₀₀ of 45.6 kg./cm.².

Example 14A

For purposes of comparison, a fluoropolymer outside the invention is made in substantially the same manner as in Example 14 except for the omission of the bromotrifluoroethylene; and the resulting copolymer of tetrafluoroethylene, vinylidene fluoride and propylene is compounded and press-cured as in Example 14. The non-useful results are as described in Example 13A.

Example 15

A fluoropolymer is prepared in substantially the same manner as in Example 10 except the bromine compound used in that Example is replaced with 4-bromoperfluorobutene-1. This bromine compound can be made as follows: The diadduct of vinylidene fluoride and bromotrichloromethane is prepared according to German Patent 949,822. Hydrolysis of the diadduct and conversion to the ethyl ester is carried out according to the procedure of Y.K. Kim, J. Org. Chem. 32,3673 (1967). Decarboxylation of the sodium salt of the ester gives 4-bromoperfluorobutene-1.

The resulting copolymer is compounded, cured and tested in substantially the same manner as described in Example 10. Useful results are obtained.

Example 16

A fluoropolymer is prepared by repeating Example 8 except the copolymer is made from 46 parts vinylidene fluoride, 51 parts hexafluoropropylene, 41 parts tetrafluoroethylene, 12 parts bromotrifluoroethylene and (for each 140 parts monomer) 600 parts water containing 0.59 part ammonium persulfate and 0.15 part sodium hydroxide. The reaction mixture pH is 2.9, residence time is 20 minutes, and the latex copolymer solids content (after 91.2% monomer conversion) is 17.7%. The copolymer contains 36% vinylidene fluoride, 31% hexafluoropropylene, 32% tetrafluoroethylene and 1% bromotrifluoroethylene (weight % of polymerized units).

A curable fluoropolymer composition is prepared in the manner described in Example 10 except the Example 16 copolymer is used, the amount of triallyl-isocyanurate is 1.5 parts and the amount of Example 9 peroxide curing agent is 2 parts. Test samples are press-cured 15 minutes at 177°C and oven-cured 4 hours to reach 260°C and 18 hours at 260°C.

The resulting cured fluoropolymer has a compression set value of 24 (pellet, 232°C for 70

hours), a T_B of 150 kg./cm.², an E_B of 185% and an M_{100} of 57.7 kg./cm.².

Example 17

A fluoropolymer is prepared by repeating Example 8 except the copolymer is made from 55 parts vinylidene fluoride, 35 parts perfluoromethyl perfluorovinyl ether, 10 parts tetrafluoroethylene and 0.5 part vinyl bromide (added as a 16% by volume solution in trichlorotrifluoroethane) and, for each 100 parts of monomer, 400 parts water containing 0.6 part ammonium persulfate and 0.1 part sodium hydroxide. The reaction mixture pH is 3.3, residence time is 30 minutes, and the latex copolymer solids content, after 97.5% monomer conversion, is 20%. The copolymer contains 55% vinylidene fluoride, 34.5% perfluoromethyl perfluorovinyl ether, 10% tetrafluoroethylene and 0.5% vinyl bromide (weight % of copolymerized units). It has a Mooney viscosity of 32 (Example 8 method).

A curable fluoropolymer composition is prepared by mixing on a two-roll mill 100 parts of the Example 17 copolymer, 30 parts carbon black MT, 1 part magnesium oxide, 3 parts dibasic lead phosphite, 5 parts of the Example 9 peroxide curing agent and 4 parts triallyl isocyanurate. Test samples are press-cured 15 minutes at 177°C and oven-cured 4 hours to reach 260°C and 24 hours at 260°C. The resulting cured fluoropolymer has a compression set value of 66 (0-rings, 70 hours at 232°C), a T_B of 105 kg./cm.² an E_B of 165% and an M_{100} of 54.2 kg./cm.².

When a fluoropolymer outside the invention is made, for purposes of comparison, in substantially the same manner as in Example 17 except for the omission of the vinyl bromide, and the resulting copolymer of vinylidene fluoride, perfluoromethyl perfluorovinyl ether and tetrafluoroethylene is compounded and press-cured for 15 minutes at 177°C, one obtains a spongy non-useful product of the type described in Example 13A.

WHAT WE CLAIM IS:

1. A fluoropolymer composition which comprises (A) a copolymer which includes interpolymerized units of
 - (1) up to 3 mole % of units derived from a bromine-containing olefin with the proviso that enough of such units are present to provide at least 0.05 weight % bromine in the copolymer, and
 - (2) enough units derived from component (2.1) or (2.2) as follows to make up the remainder of the copolymer:
 - (2.1) at least one fluorine-containing monomer which is copolymerizable with component (1), contains 2-7 carbon atoms, contains no bromine atom, and contains at least as many fluorine atoms as carbon atoms, or
 - (2.2) 50-99 mole % of tetrafluoroethylene, 1-50 mole % of a C₂-C₄ olefinic hydrocarbon and from 0 to 30 mole % of vinylidene fluoride, and
- (B) 0.5-10% of an organic peroxide based on the weight of component (A).
2. A composition as claimed in Claim 1 wherein component (B) is a dialkyl peroxide.
3. A composition as claimed in Claim 1 or Claim 2 which additionally contains
- (C) 1-15%, based on the weight of component (A), of a metal compound selected from divalent metal oxides and divalent metal hydroxides.
4. A composition as claimed in Claim 3 which also contains
- (D) 0.5-10%, based on the weight of component (A) of a co-agent composed of a polyunsaturated compound which is capable of cooperating with said peroxide to provide a useful cure.
5. A composition as claimed in Claim 4 wherein component (D) is a compound selected from triallyl cyanurate; triallyl isocyanurate; tris(diallylamine)-s-triazine; triallyl phosphite; N,N-diallyl acrylamide; hexa-allyl phosphoramidate; N,N,N',N'-tetra-allyl terephthalamide; N,N,N',N'-tetra-allyl malonamide; trivinyl isocyanurate; 2,4,6-trivinyl methyltrisiloxane; and tri(5-norbornene-2-methylene)cyanurate.
6. A composition as claimed in any of claims 1-5 wherein the units of component (A-1) are derived from bromotrifluoroethylene and/or 4-bromo-3,3,4,4-tetrafluorobutene-1.
7. A composition as claimed in any of Claims 1-5 wherein component (A-1) is 1-bromo-2,2-difluoroethylene.
8. A composition as claimed in any of Claims 1-7 wherein enough component (A-1) units are present to provide 0.3-1.5 weight % bromine in the copolymer.
9. A composition as claimed in any of Claims 1-8 wherein component (A-2) is composed of units derived from tetrafluoroethylene and a perfluoroalkyl perfluorovinyl ether wherein the perfluoroalkyl group contains 1-5 carbon atoms.
10. A composition as claimed in Claim 9 wherein said ether is perfluoromethyl perfluorovinyl ether.
11. A composition as claimed in Claim 9 or Claim 10 wherein the tetrafluoroethylene content of (A-2) is 50-80 mole % and the ether content is 20-50 mole %.
12. A composition as claimed in any of Claims 1-8 wherein component (A-2) is

composed of units derived from vinylidene fluoride, at least one fluorine-containing olefin copolymerizable therewith, and optionally a perfluoroalkyl perfluorovinyl ether wherein the perfluoroalkyl group contains 1-5 carbon atoms.

13. A composition as claimed in Claim 12 wherein component (A-2) is composed of units derived from any one of the following three combinations:

- (i) vinylidene fluoride and hexafluoropropylene or pentafluoropropylene; or
- (ii) vinylidene fluoride, tetrafluoroethylene and hexafluoropropylene or pentafluoropropylene; or
- (iii) vinylidene fluoride, a perfluoroalkyl perfluorovinyl ether wherein the perfluoroalkyl group contains 1-5 carbon atoms, and hexafluoropropylene and/or tetrafluoroethylene.

14. A composition as claimed in Claim 12 or Claim 13 wherein component (A-2) is composed of units derived from 30-70% vinylidene fluoride and 30-70% hexafluoropropylene based on the weight of A-2.

15. A composition as claimed in any of claims 12-14 wherein component (A-2) is composed of units derived from 25-70% vinylidene fluoride, 19-60% hexafluoropropylene and 3-35% tetrafluoroethylene based on the weight of A-2.

16. A composition as claimed in any of Claims 1-8 wherein component (A-2) is composed of units derived from tetrafluoroethylene alone or units from 50-99 mole % tetrafluoroethylene and 1-50 mole % propylene.

17. A composition as claimed in any of Claims 1-8 wherein component (A-2) is composed of units derived from vinylidene fluoride alone or units from 1-30 mole % vinylidene fluoride, 50-98 mole % tetrafluoroethylene and 1-48 mole % propylene.

18. A composition as claimed in any of the preceding claims wherein component (B) is a peroxide which decomposes at a temperature above 49°C.

19. A composition as claimed in any of the preceding claims wherein component (B) is a di-tertiarybutyl peroxide.

20. A composition as claimed in any of Claims 1-18 wherein component (B) is 2,5-dimethyl-2,5-di(tertiarybutylperoxy) hexyne-3.

21. A composition as claimed in any of Claims 1-18 wherein component (B) is 2,5-dimethyl-2,5-di(tertiarybutylperoxy) hexane.

22. A composition as claimed in any of the preceding claims wherein the component (B) content is 1.5-5% based on the weight of component (A).

23. A composition as claimed in any of Claims 3-22 wherein the component (C) content is 2-10% based on the weight of component (A).

24. A composition as claimed in any one of Claims 4-23 wherein the component (D) content is 1-7% based on the weight of component (A).

25. A composition as claimed in Claim 24 wherein component (D) is triallyl isocyanurate.

26. A composition as claimed in Claim 1 substantially as hereinbefore described.

27. A composition as claimed in Claim 1 substantially as hereinbefore described with reference to the Examples.

28. An article whenever prepared by curing a composition as claimed in Claim 1.

Agents for the Applicants

FRANK B. DEHN & CO.,
Imperial House,
15-19, Kingsway,
LONDON, WC2B 6LA.